1. Problem 17-2 from McQuarrie & Simon:

Show that Equation 17.8 is equivalent to f(x+y) = f(x)f(y). In this problem, we will prove that $f(x) \propto e^{ax}$. First, take the logarithm of the above equation to obtain

$$\ln f(x+y) = \ln f(x) + \ln f(y)$$

Differentiate both sides with respect to x (keeping y fixed) to get

$$\left[\frac{\partial \ln f(x+y)}{\partial x}\right]_{y} = \frac{d \ln f(x+y)}{d(x+y)} \left[\frac{\partial (x+y)}{\partial x}\right]_{y} = \frac{d \ln f(x+y)}{d(x+y)}$$
$$= \frac{d \ln f(x)}{dx}$$

Now differentiate with respect to y (keeping x fixed) and show that

$$\frac{d\ln f(x)}{dx} = \frac{d\ln f(y)}{dy}$$

For this relation to be true for all x and y, each side must equal a constant, say a. Show that

$$f(x) \propto e^{ax}$$
 and $f(y) \propto e^{ay}$

2. Problem 17-5 from McQuarrie & Simon:

Show that the partition function in Example 17-1 can be written as

$$Q(\beta, B_z) = 2 \cosh\left(\frac{\beta \hbar \gamma B_z}{2}\right) = 2 \cosh\left(\frac{\hbar \gamma B_z}{2k_B T}\right)$$

Use the fact that $d \cosh x / dx = \sinh x$ to show that

$$\langle E \rangle = -\frac{\hbar \gamma B_z}{2} \tanh \frac{\beta \, \hbar \gamma B_z}{2} = -\frac{\hbar \gamma B_z}{2} \tanh \frac{\hbar \gamma B_z}{2k_BT}$$

3. Problem 17-6 from McQuarrie & Simon:

Use either the expression for $\langle E \rangle$ in Example 17–1 or the one in Problem 17–5 to show that

$$\langle E \rangle \to -\frac{\hbar \gamma B_z}{2}$$
 as $T \to 0$

and that

$$\langle E \rangle \to 0$$
 as $T \to \infty$

4. Problem 17-9 from McQuarrie & Simon:

In Section 17–3, we derived an expression for $\langle E \rangle$ for a monatomic ideal gas by applying Equation 17.20 to Q(N, V, T) given by Equation 17.22. Apply Equation 17.21 to

$$Q(N, V, T) = \frac{1}{N!} \left(\frac{2\pi m k_B T}{h^2} \right)^{3N/2} V^N$$

to derive the same result. Note that this expression for Q(N, V, T) is simply Equation 17.22 with β replaced by $1/k_BT$.

5. Problem 17-11 from McQuarrie & Simon:

Although we will not do so in this book, it is possible to derive the partition function for a monatomic van der Waals gas.

$$Q(N, V, T) = \frac{1}{N!} \left(\frac{2\pi m k_B T}{h^2} \right)^{3N/2} (V - Nb)^N e^{aN^2/V k_B T}$$

where a and b are the van der Waals constants. Derive an expression for the energy of a monatomic van der Waals gas.

6. Problem 17-20 from McQuarrie & Simon:

Deriving the partition function for an Einstein crystal is not difficult (see Example 17-3). Each of the N atoms of the crystal is assumed to vibrate independently about its lattice position, so that the crystal is pictured as N independent harmonic oscillators, each vibrating in three directions. The partition function of a harmonic oscillator is

$$q_{\text{ho}}(T) = \sum_{v=0}^{\infty} e^{-\beta \left(v + \frac{1}{2}\right)hv}$$
$$= e^{-\beta hv/2} \sum_{v=0}^{\infty} e^{-\beta vhv}$$

This summation is easy to evaluate if you recognize it as the so-called geometric series (MathChapter I)

$$\sum_{v=0}^{\infty} x^v = \frac{1}{1-x}$$

Show that

$$q_{\text{ho}}(T) = \frac{e^{-\beta h \nu/2}}{1 - e^{-\beta h \nu}}$$

and that

$$Q = e^{-\beta U_0} \left(\frac{e^{-\beta h\nu/2}}{1 - e^{-\beta h\nu}} \right)^{3N}$$

where U_0 simply represents the zero-of-energy, where all N atoms are infinitely separated.

Numbered equations:

$$f(E_1 - E_3) = f(E_1 - E_2) f(E_2 - E_3)$$
(17.8)

$$\langle E \rangle = -\left(\frac{\partial \ln Q}{\partial \beta}\right)_{NV} \tag{17.20}$$

$$\langle E \rangle = k_B T^2 \left(\frac{\partial \ln Q}{\partial \beta} \right)_{NV} \tag{17.21}$$

$$Q(N, V, \beta) = \frac{[q(V, \beta)]^{N}}{N!}$$
(17.22)

From Example 17–1:

$$Q(T,B_z) = e^{\beta \hbar \gamma B_z/2} + e^{-\beta \hbar \gamma B_z/2} = e^{\hbar \gamma B_z/2k_BT} + e^{-\hbar \gamma B_z/2k_BT}$$

$$\begin{split} \langle E \rangle &= - \left(\frac{\partial \ln Q}{\partial \beta} \right)_{B_z} = - \frac{1}{Q (\beta, B_z)} \left(\frac{\partial Q}{\partial \beta} \right)_{B_z} \\ &= - \frac{\hbar \gamma B_z}{2} \left(\frac{e^{\beta \hbar \gamma B_z/2} - e^{-\beta \hbar \gamma B_z/2}}{e^{\beta \hbar \gamma B_z/2} + e^{-\beta \hbar \gamma B_z/2}} \right) \\ &= - \frac{\hbar \gamma B_z}{2} \left(\frac{e^{\hbar \gamma B_z/2k_BT} - e^{-\hbar \gamma B_z/2k_BT}}{e^{\hbar \gamma B_z/2k_BT} + e^{-\hbar \gamma B_z/2k_BT}} \right) \end{split}$$