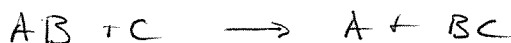


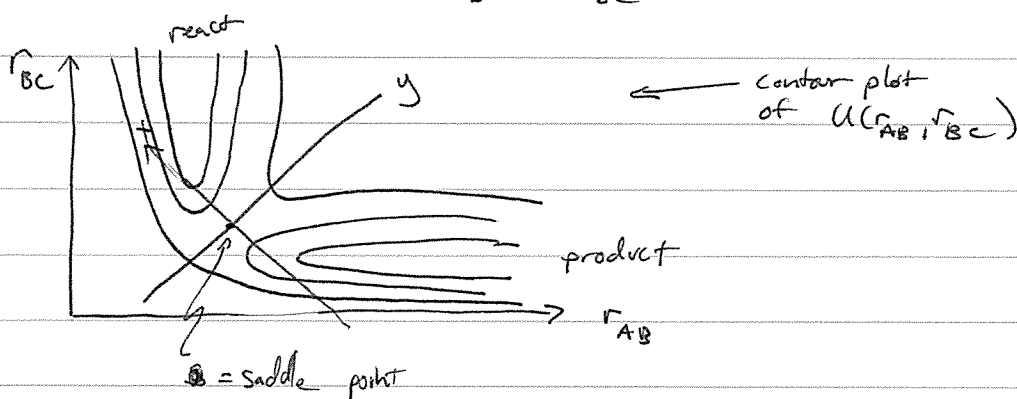
36-1

Transition state theory:

consider a reaction:



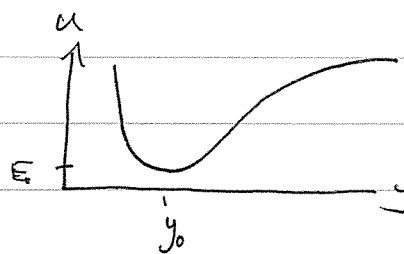
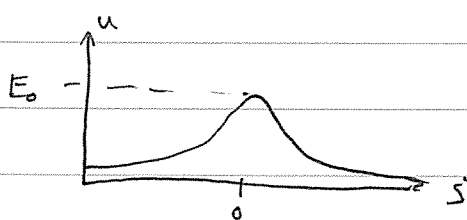
(lying in a line:



$$\begin{aligned} s &= r_{BC} - r_{AB} \\ y &= r_{BC} + r_{AB} \end{aligned} \quad \left. \vphantom{\begin{aligned} s &= r_{BC} - r_{AB} \\ y &= r_{BC} + r_{AB} \end{aligned}} \right\} \begin{array}{l} \text{new coordinates at} \\ \text{saddle point.} \end{array}$$

s = reaction path

Consider potential along these 2 curves



At the saddle point,

$$\frac{\partial U}{\partial s} = 0$$

$$\frac{\partial^2 U}{\partial s^2} < 0$$

imaginary frequency at saddle

$$\frac{\partial U}{\partial y} = 0$$

$$\frac{\partial^2 U}{\partial y^2} > 0$$

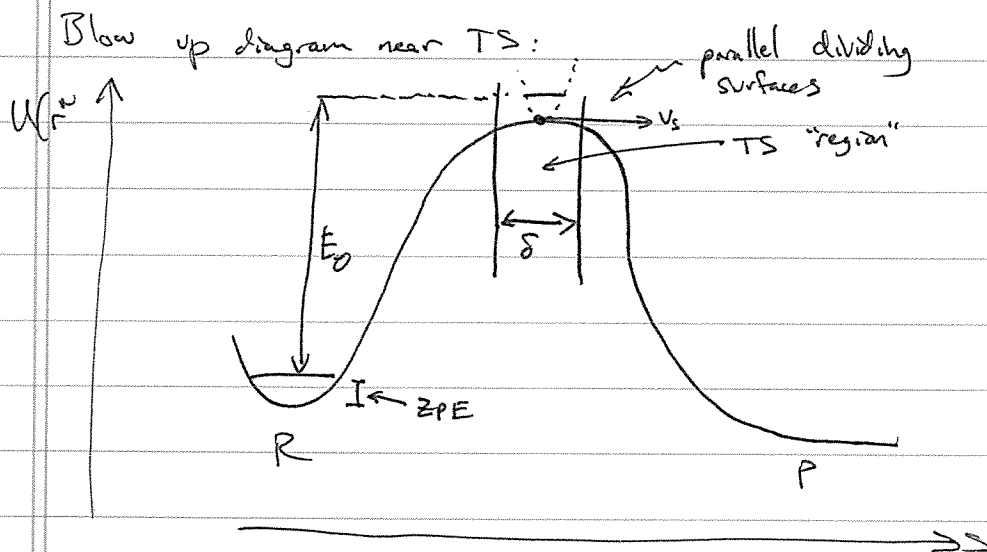
real frequency at saddle

Approximate surface ~~at~~ ^{near} saddle:

$$U(s, y) = \frac{1}{2} G_y (y - y_0)^2 - \frac{1}{2} G_s s^2 + E_0$$

Basic tenets of TST

- 1) molecules that have crossed the TS in the direction of products can't turn around to form reactants
- 2) At the TS, motion along reaction coordinate is separable from other motions and is treated as a translation.
- 3) TS is in a state of quasi-equilibrium and equilibrium distributions apply. (actually follows from point 1)



Molecules ~~moving~~ moving forward out of TS region: $N_f^\ddagger$

Molecules moving backwards out of TS region: $N_b^\ddagger$

$$N^\ddagger = N_f^\ddagger + N_b^\ddagger \quad \text{and} \quad K^\ddagger = \frac{N^\ddagger}{[AB][C]} = \frac{N_f^\ddagger}{N_b^\ddagger}$$

or

$$N^\ddagger = K^\ddagger [AB][C]$$

Now, let's consider only those forward moving states while we're still at eqm with reactants:

$$\text{eqm.} \quad N_f^\ddagger = N_b^\ddagger = N^\ddagger / 2$$

$$\text{so:} \quad N_f^\ddagger = K^\ddagger [AB][C] / 2$$

← quasi-equilibrium hypothesis

Net rate of reaction in forward direction

$$\frac{dN}{dt} (r \rightarrow p) = \frac{\delta N^\ddagger}{\delta t}$$

Now if we know the average velocity $\bar{v}_s$ along RC,

$$\bar{v}_s = \frac{\delta}{\delta t} \quad \leftarrow \text{gap between dividing surfaces}$$

$$\delta t = \delta / \bar{v}_s$$

So:

$$\frac{dN}{dt} = \delta N^\ddagger \frac{\bar{v}_s}{\delta} \quad \leftarrow \text{but this is just half the total population in the TS region}$$

$$\delta N^\ddagger = \frac{N^\ddagger}{2}$$

$$\therefore \frac{dN}{dt} = \frac{N^\ddagger}{2} \frac{\bar{v}_s}{\delta} \quad \leftarrow \text{all we need is this!}$$

$$\bar{v}_s = \frac{\int_0^\infty v_s e^{-(\mu_s v_s^2 / k_B T)} dv_s}{\int_0^\infty e^{-\mu_s v_s^2 / k_B T} dv_s} \quad \leftarrow \text{equilibrium distribution of velocities at TS}$$

we only consider forward moving TS

$$\mu_s = \text{reduced mass of reaction coord} = \frac{m_{AB} m_c}{m_{AB} + m_c}$$

$$\bar{v}_s = \sqrt{\frac{2k_B T}{\pi \mu_s}}$$

$$\therefore \frac{dN}{dt} = \frac{N^\ddagger}{2} \left(\frac{2k_B T}{\pi \mu_s} \right)^{1/2} \frac{1}{\delta}$$

Now remember that:

$$K_c^\ddagger = \frac{N^\ddagger}{[AB][C]} = \frac{Q_{tot}^\ddagger}{Q_{AB} Q_C} e^{-E_0 / k_B T} \quad \leftarrow \text{total TPF at TS}$$

36-4

$$\frac{dN}{dt} = \sqrt{\frac{2kT}{\pi\mu_s}} \frac{1}{2\delta} \frac{Q_{tot}^\ddagger}{Q_A Q_C} e^{-E_0/k_B T} [AB][C]$$

$$Q_{tot}^\ddagger = Q_s Q^\ddagger = Q^\ddagger \sqrt{\frac{2\pi\mu_s k_B T}{h^2}} \delta/h$$

$\uparrow$ 1-D translational PF along s $\nwarrow$ 1-D box size

$$\frac{dN}{dt} = \frac{k_B T}{h} \frac{Q^\ddagger}{Q_A Q_C} e^{-E_0/k_B T} [AB][C]$$

Consider 1st order rate law

$$\frac{dN}{dt} = k [AB][C]$$

rate const

$$k = \frac{k_B T}{h} \frac{Q^\ddagger}{Q_A Q_C} e^{-E_0/k_B T}$$

Everything artificial δ, μ_s, v_s , factors of $\frac{1}{2}$ have cancelled!

$Q^\ddagger$ is the PF for the activated complex with relative motion (s) frozen.

$A \cdots B \cdots C \longrightarrow$ still has 3 translations
still has 3 rotations (but geometry is different)

has $3N - 6 - 1$ vibrations

$\uparrow$ asymmetric stretch is now s! and has (imaginary frequency)